



## An unexpected rearrangement of the benzofurobenzazepine skeleton of galanthamine-type alkaloids

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### ABSTRACT

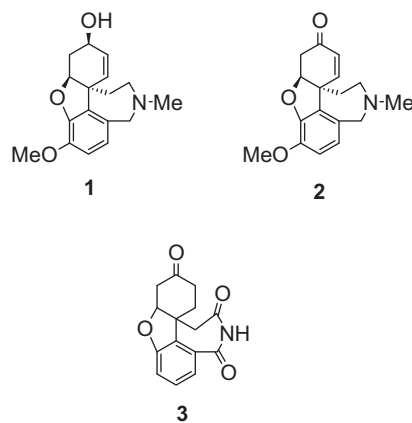
Attempted cyclisation of *N*-methylated spiro benzazepine–cyclohexenone (**5**) into the corresponding *N*-methyl tetracyclic unit of galanthamine-type alkaloids (**6**) instead gave an unexpected rearrangement to yield a cyclopentanoisoquinolinone derivative (**7**). Methylation of the tetrahydrobenzofurobenzazepine tetracycle resulted in the expected *N*-methyl derivative **6**, and the anomalous product **8**, with structure similar to that of **7**.

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(–)-Galanthamine (**1**) is an *Amaryllidaceae* alkaloid,<sup>1</sup> isolated from the flowers and bulbs of the Caucasian snowdrop (*Galanthus woronowii*), and exhibits competitive and reversible acetylcholine esterase (AChE) inhibition. Moreover, this molecule displays allosteric potentiation of neuronal nicotinic receptors for acetylcholine.<sup>2</sup> (–)-Galanthamine hydrobromide (Razadyne, Reminyl) enhances significantly cognitive functions and is used for the treatment of mild to moderate Alzheimer's disease.<sup>3,4</sup>

A number of synthetic routes have been elaborated<sup>1</sup> for the preparation of (–)-galanthamine using different key steps to form the tetracyclic ring system characteristic of galanthamine-type *Amaryllidaceae* alkaloids.<sup>5–9</sup> Most synthetic strategies utilized a biomimetic approach via intramolecular phenolic oxidative coupling to install the quaternary spiro carbon. In relation to this type of synthetic processes, narwedine (**2**) as well as its biogenetic precursor<sup>10</sup> can be considered the most important intermediates.

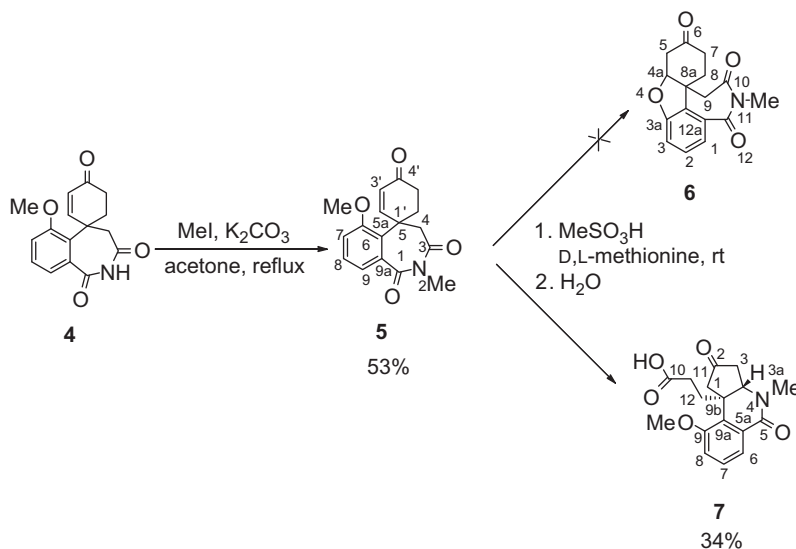
Recently,<sup>11</sup> we succeeded in synthesizing the hexahydrobenzofurobenzazepine trione **3** which represents the appropriate key intermediate for the preparation of the demethoxy derivative of narwedine (**2**). The synthesis of compound **3** was achieved starting from spirocyclohexenone derivative **4** in one step via demethylation of the methoxy group and cyclisation using methanesulfonic acid in the presence of methionine.<sup>11</sup>



The spiro compound **4**<sup>11,12</sup> was *N*-methylated with methyl iodide in refluxing acetone in the presence of anhydrous potassium carbonate (Scheme 1) giving *N*-methyl derivative **5** in 53% yield. Next, the cyclisation reaction was performed like that used for the preparation of tetracycle **3**. Compound **5** was allowed to react in methanesulfonic acid in the presence of racemic methionine at room temperature for several hours. After work-up, however, the *N*-methyl tetracycle **6** was not isolated, but instead the unexpected cyclopentanoisoquinolinone derivative **7** was obtained. This type of tricyclic dione is unknown in the literature.

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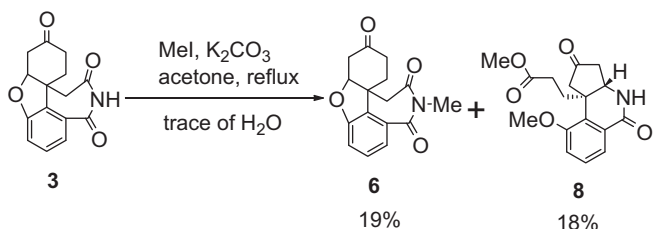
E-mail address: [szantay@mail.bme.hu](mailto:szantay@mail.bme.hu) (C. Szántay).



Scheme 1.

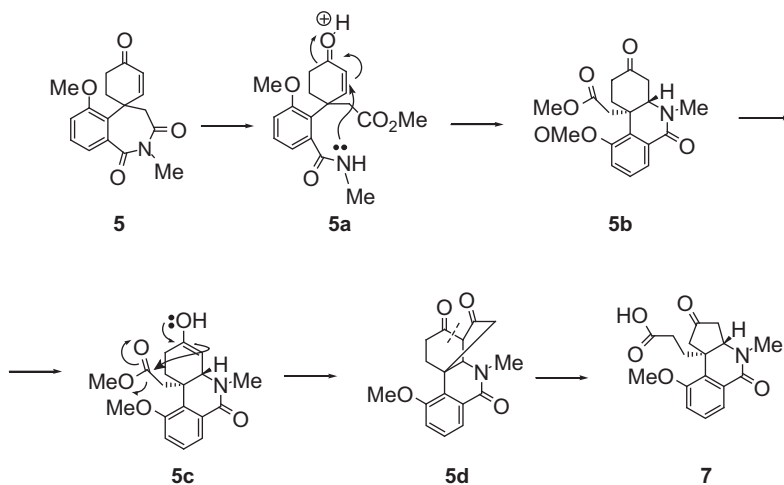
Our next attempt to obtain the target molecule was to allow the hexahydrobenzofurobenzazepine tetracycle **3** to react with methyl iodide so as to prepare the *N*-methyl imide **6**. The reaction conditions in this procedure were the same as those used for the methylation of spiro derivative **4** (Scheme 2).

In this reaction two products were isolated: the expected *N*-methyl tetracycle **6** was obtained in 19% yield, and surprisingly, as in the previously attempted cyclisation reaction, the analogous tricyclic cyclopentanoisoquinoline dione **8** was also obtained.

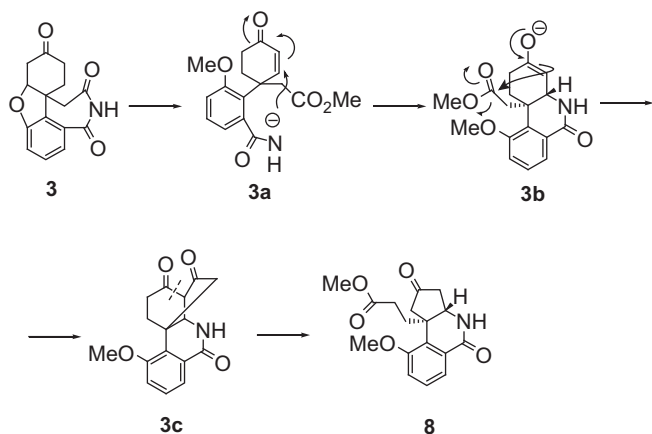


Scheme 2.

High-resolution mass spectrometric data indicated that compound **7**<sup>13</sup> had the elementary composition C<sub>17</sub>H<sub>19</sub>NO<sub>5</sub> (as opposed to C<sub>16</sub>H<sub>15</sub>NO<sub>4</sub> for compound **6**). In the <sup>1</sup>H NMR spectra, one of the most conspicuous differences between **7** and **6** is a doublet of doublets signal belonging to a CH methine proton at 4.10 ppm in the <sup>1</sup>H spectrum of **7** instead of the expected triplet signal at 4.90 ppm due to H-4a as for **6**. The corresponding carbon signal in **7** occurred at 62.7 ppm, (86.5 ppm in **6**). These chemical shifts indicate that this methine is situated next to a nitrogen instead of an oxygen atom. In addition, the –CH<sub>2</sub>–C(O)–CH<sub>2</sub>– ketone carbon signal was shifted downfield by 4.5 ppm in the <sup>13</sup>C spectrum of **7** compared to **6**, suggesting that this carbonyl group is in a strained five-membered ring. There was a singlet at 3.84 ppm in the <sup>1</sup>H spectrum of **7** due to an additional *O*-methyl group connected to the aromatic ring. A broad singlet at 12.0 ppm, not connected to a carbon, was likely to belong to a carboxylic acid OH. From a structurally informative point of view, the 2D <sup>1</sup>H–<sup>1</sup>H correlation spectrum showed a connection between the CH signal at 4.10 ppm (H-3a) and the diastereotopic CH<sub>2</sub> signals at 1.94 and 2.86 ppm (CH<sub>2</sub>-3), and also between the CH<sub>2</sub> signals at 1.94–1.98 and 2.13–2.18 ppm (CH<sub>2</sub>-11) and 2.00–2.05 and 2.22–2.26 ppm (CH<sub>2</sub>-12). These data, together with further <sup>1</sup>H–<sup>1</sup>H, direct <sup>1</sup>H–<sup>13</sup>C, and long-range



Scheme 3.



Scheme 4.

$^1\text{H}$ – $^{13}\text{C}$  scalar connectivities as measured from 2D experiments, allowed the unambiguous determination of the structure of **7** as shown in Scheme 2. The  $^1\text{H}$  and  $^{13}\text{C}$  assignments derived from, and supported by, the 2D scalar correlation experiments are provided.<sup>13</sup> The relative configuration of C(3a) and C(9b) follows from the fact that the measured vicinal  $J_{3a,3\beta} = 8.1$  Hz and  $J_{3a,3\alpha} = 10.6$  Hz couplings are only consistent with the *trans* geometry of the ring fusion.

The structural elucidation of compound **8**<sup>14</sup> (Scheme 2) rests on an entirely analogous procedure and argument as discussed for compound **7**.

The rearrangement yielding compounds **7** and **8** may be plausibly rationalized in terms of the mechanisms shown for an acidic medium in Scheme 3, and for a basic medium in Scheme 4.

In the latter case, cleavage of the ether bond provides a phenol which is subsequently methylated, and in addition a cyclohexenone is formed. After conjugate addition of the nitrogen atom on the enone, Claisen condensation occurs between the cyclohexenone methylene group and the ester group. Subsequently, a retro-Claisen condensation forms the cyclopentanone ring and the side chain, yielding the cyclopentanoisoquinoline **8**.

To sum up, we may conclude that in the course of an attempted cyclisation of compound (**5**) into a unit of galanthamine-type alkaloids (**6**) an unexpected rearrangement occurred to yield a new derivative (**7**). Subsequent methylation resulted in the expected derivative **6**, and the anomalous product **8**.

## Acknowledgements

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- Spectral data are as follows for compound 7*. IR (KBr): 3436, 2929, 1747, 1641, 1277, 1256, 820, 755  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR, (800 MHz,  $\text{DMSO}-d_6$ ): 1.94 (ddd, 1H,  $J_{gem} = 18.6$  Hz,  $J_{3\alpha,3a} = 10.6$  Hz,  $J_{3\alpha,1\alpha} = 2.2$  Hz,  $\text{H}_{\alpha-3}$ ), 1.94–1.98 (m, 1H,  $\text{H}_{\alpha-11}$ ), 2.00–2.05 (m, 1H,  $\text{H}_{\alpha-12}$ ), 2.13–2.18 (m, 1H,  $\text{H}_{\gamma-11}$ ), 2.22–2.26 (m, 1H,  $\text{H}_{\gamma-12}$ ), 2.60 (dd, 1H,  $J_{gem} = 18.5$  Hz,  $J_{1\beta,3\beta} = 1.3$  Hz,  $\text{H}_{\beta-1}$ ), 2.86 (ddt, 1H,  $J_{gem} = 18.6$  Hz,  $J_{3\beta,3a} = 8.1$  Hz,  $J_{3\beta,1\alpha} = 3.8$  Hz,  $J_{3\beta,1\beta} = 1.3$  Hz,  $\text{H}_{\beta-3}$ ), 3.57 (ddd, 1H,  $J_{gem} = 18.5$  Hz,  $J_{1\alpha,3\alpha} = 2.2$  Hz,  $J_{1\alpha,3\beta} = 1.3$  Hz,  $\text{H}_{\alpha-1}$ ), 3.04 (s, 3H,  $\text{NCH}_3$ ), 3.84 (s, 3H,  $\text{CH}_3\text{O}$ ), 4.10 (dd, 1H,  $J_{3a,3\alpha} = 10.6$  Hz,  $J_{3a,3\beta} = 8.1$  Hz,  $\text{H}_{\beta-3a}$ ), 7.26 (dd, 1H,  $J_{8,7} = 8.3$  Hz,  $J_{8,6} = 1.2$  Hz,  $\text{H}-8$ ), 7.40 (dd, 1H,  $J_{7,8} = 8.3$  Hz,  $J_{7,6} = 7.8$  Hz,  $\text{H}-7$ ), 7.64 (dd, 1H,  $J_{6,7} = 7.8$  Hz,  $J_{6,8} = 1.2$  Hz,  $\text{H}-6$ ), 12.0 (br s, 1H,  $\text{H}-\text{COOH}$ ).  $^{13}\text{C}$  NMR (200 MHz,  $\text{DMSO}-d_6$ ): 29.8 (C-11), 33.8 (C-12), 33.9 ( $\text{NCH}_3$ ), 43.1 (C-3), 44.4 (C-9b); 49.5 (C-1), 55.7 ( $\text{OCH}_3$ ), 62.7 (C-3a), 115.9 (C-8), 120.3 (C-6), 126.6 (C-9a), 128.5 (C-7), 130.2 (C-5a), 157.3 (C-9), 161.6 (C-5), 174.0 (C-10), 212.6 (C-2) ppm. MS (ESI):  $M^+$ :  $m/z$  318.13320, calcd value for  $\text{C}_{17}\text{H}_{20}\text{NO}_5$ : 318.13360 ( $\Delta$ :  $-1.2$  ppm).  $\text{MS}^2$  of  $m/z$  318:  $m/z$  300 (100), 258 (2).
- Spectral data are as follows for compound 8*. IR (KBr): 3409, 1745, 1733, 1671, 1261, 1053, 758  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (800 MHz,  $\text{DMSO}-d_6$ ): 2.01–2.10 (m, 3H,  $\text{H}_{\alpha-11}$ ,  $\text{H}_{\alpha-12}$ ,  $\text{H}_{\alpha-3}$ ), 2.26–2.30 (m, 1H,  $\text{H}_{\gamma-12}$ ), 2.35–2.39 (m, 1H,  $\text{H}_{\gamma-11}$ ), 2.58 (dd, 1H,  $J_{gem} = 18.5$  Hz,  $J_{1\beta,3\beta} = 1.2$  Hz,  $\text{H}_{\beta-1}$ ), 2.62 (ddt, 1H,  $J_{gem} = 18.5$  Hz,  $J_{3\beta,3a} = 8.0$  Hz,  $J_{3\beta,1\beta} = 1.2$  Hz,  $\text{H}_{\beta-3}$ ), 3.46 (dd, 1H,  $J_{gem} = 18.5$  Hz,  $J_{1\alpha,3\alpha} = 2.0$  Hz,  $\text{H}_{\alpha-1}$ ), 3.51 (s, 3H, C(10)- $\text{OCH}_3$ ), 3.83 (s, 3H, C(9)- $\text{OCH}_3$ ), 4.02 (ddd, 1H,  $J_{3a,3\alpha} = 10.2$  Hz,  $J_{3a,3\beta} = 8.0$  Hz,  $J_{3a,4} = 4.9$  Hz,  $\text{H}_{\beta-3a}$ ), 7.25 (dd, 1H,  $J_{8,7} = 8.3$  Hz,  $J_{8,6} = 1.2$  Hz,  $\text{H}-8$ ), 7.39 (dd, 1H,  $J_{7,8} = 8.3$  Hz,  $J_{7,6} = 7.7$  Hz,  $\text{H}-7$ ), 7.61 (dd, 1H,  $J_{6,7} = 7.7$  Hz,  $J_{6,8} = 1.2$  Hz,  $\text{H}-6$ ), 8.28 (d, 1H,  $J_{4,3a} = 4.9$  Hz,  $\text{H}-4$ ).  $^{13}\text{C}$  NMR (200 MHz,  $\text{DMSO}-d_6$ ): 29.7 (C-11), 33.3 (C-12), 44.1 (C-9b), 45.4 (C-3), 49.2 (C-1), 51.3 (C(10)- $\text{OCH}_3$ ), 55.0 (C-3a), 55.7 (C(9)- $\text{OCH}_3$ ), 116.1 (C-8), 120.1 (C-6), 127.3 (C-9a), 128.4 (C-7), 130.2 (C-5a), 157.5 (C-9), 162.9 (C-5), 172.9 (C-10), 213.2 (C-2) ppm. MS (ESI):  $M^+$ :  $m/z$  318.13320, calcd value for  $\text{C}_{17}\text{H}_{20}\text{NO}_5$ : 318.13360 ( $\Delta$ :  $-1.3$  ppm).  $\text{MS}^2$  of  $m/z$  318:  $m/z$  286 (100), 244 (4).